

## What Is A Supersaturated Solution In Science

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### What Is A Supersaturated Solution

A supersaturated solution is a more solute solution than can be dissolved by the solvent. If you haven't learned what a solute / solvent is, the material that is dissolved in the solution, such as salts but not restricted to salts, is a solution.

### Supersaturated Solution - Definition, Examples ...

A supersaturated solution is a solution with more dissolved solute than the solvent would normally dissolve in its current conditions. Supersaturation is achieved by dissolving a solute in one set of conditions, then transferring it to other conditions without triggering any release of the solute.

### What Is a Supersaturated Solution?

A supersaturated solution is a solution that contains more than the maximum amount of solute that is capable of being dissolved at a given temperature. The recrystallization of the excess dissolved solute in a supersaturated solution can be initiated by the addition of a tiny crystal of solute, called a seed crystal.

### Supersaturated Solutions | Chemistry for Non-Majors

Supersaturated solutions are very unstable, and the solute will readily fall out of solution if disturbed. As it does this, crystals can form, releasing heat in the process. Making a ...

### Supersaturated Solution: Definition & Example - Video ...

Supersaturation occurs with a chemical solution when the concentration of a solute exceeds the concentration specified by the value equilibrium solubility.Most commonly the term is applied to a solution of a solid in a liquid. A supersaturated solution is in a metastable state; it may be brought to equilibrium by forcing the excess of solute to separate from the solution.

### Supersaturation - Wikipedia

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### OneClass: What is a supersaturated solution?

A super-saturated solution of glucose in boiling water crystallizes as it cools. Reason The solubility increases as the temperature decreases. MEDIUM. View Answer. A beaker contains a solution of substance ' A '. On dissolving substance ' A ' in small amount in this solution, precipitation of substance ' A ' takes place.

### What is meant by supersaturated solution?

A solution which has reached saturation at a higher temperature is supersaturated. Solution containing more solute than the slovent would originally dissolve, usually done by heating up the solution.

### What is a supersaturated solution? - Answers

A solution can be one of either gases or liquids. A solution is simply a mixture of a solute and a solvent. A solute is matter that is dissolvable. A solvent is the gas or liquid in which the solute dissolves. There are three types of solutions: unsaturated, saturated, and supersaturated. Below is a look at each of these solutions and their ...

### Unsaturated vs Saturated vs Supersaturated solutions ...

What Is a Supersaturated Solution? The definition of a supersaturated solution is one which contains more dissolved solute than could ordinarily dissolve into the solvent. A minor disturbance of the solution or introduction of a "seed" or tiny crystal of solute will force crystallization of excess solute. One way supersaturation can occur is by ...

### Saturated Solution Definition and Examples

solution [so-loo' shun] 1. a homogeneous mixture of one or more substances (solute)s dispersed molecularly in a sufficient quantity of dissolving medium (solvent). 2. In pharmacology, a liquid preparation of one or more soluble chemical substances, which are usually dissolved in water. For names of specific solutions, see under the name. 3. the process ...

### Supersaturated solution | definition of supersaturated ...

Supersaturated Solution Definition: A solution with more dissolved solute at a given temperature, than a saturated solution.

### What is a Supersaturated Solution? - Definition by Game Smartz

Supersaturated solutions are formed if you add additional solute to the saturated solution. In other words, it is the condition in a saturated solution, when you add some additional quantity of solute into the solution.

### Difference Between Saturated and Supersaturated Solution ...

In a supersaturated solution, both cations and anions are in constant motion and move in and out of the influence scopes of other ions or molecules.Ions with opposite electrical charges are attracted to form clusters. The cluster is not a stable stage until it grows large enough to become a crystallite. The process of forming a cluster is known as nucleation, which is the initial formation ...

### Supersaturated Solution - an overview | ScienceDirect Topics

A supersaturated solution is produced when a substance — a solute — is dissolved in water or another solvent to a degree not ordinarily possible. Most substances form a supersaturated solution only with difficulty. A supersaturated solution is generally achieved through changing the conditions of a saturated solution.

### What is a Supersaturated Solution? - wiseGEEK

A solution of this composition is also described as a saturated solution since it can accommodate no more KCl. Under some circumstances it is possible to prepare a solution which behaves anomalously and contains more solute than a saturated solution. Such a solution is said to be supersaturated.

### 10.16: Saturated and Supersaturated Solutions - Chemistry ...

Define supersaturated solution, supersaturated solution synonyms, supersaturated solution pronunciation, supersaturated solution translation, English dictionary definition of supersaturated solution. A solution which contains a higher concentration of solute than does a saturated solution at that temperature.

### Supersaturated solution - definition of supersaturated ...

Define supersaturated, supersaturated synonyms, supersaturated pronunciation, ... To cause (a chemical solution) to contain more dissolved solute than is normally possible under given conditions of temperature and pressure. 2. To cause (a vapor) ...