

Review Chapter 5 The Periodic Table

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Review Chapter 5 The Periodic

CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered.

5 The Periodic Law

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Modern Chemistry 5 The Periodic Law CHAPTER 5 REVIEW The Periodic Law SECTION 2 SHORT ANSWER Use this periodic table to answer the following questions in the space provided. 1. Identify the element and write the noble-gas notation for each of the following: a. the Group 14 element in Period 4

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The left side of the periodic table is composed of metals, which have few valence electrons. Moving to the right, there are more electrons and their mobility is less, reducing their metallic properties. When most of the orbitals are filled, the atoms take on nonmetallic characteristics. When orbitals are full, a noble gas configuration is attained.

Answers to Topic 5 Periodic Table from Review Book

5.2 flashcards. The periodic trends in the periodic table are the result of ____ electron arrangement. ion, an atom, radical or molecule that has gained or lost one or more electrons and has a negative or positive charge

Chapter 5

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Chapter 5 Review 1.A horizontal row in the periodic table is called a(n) a. family. b. group. c. octet. d. period. 2. Which of the following elements is most similar in behavior to calcium? a. magnesium b. sodium c. sulfur d. Chlorine 3. The electron configurations of main-group elements end in a. d and f orbitals. b.

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Chapter 5 Review The Periodic Law Mixed Answer Key

Chapter 5 Review 1.A horizontal row in the periodic table is called a(n) a. family. b. group. c. octet. d. period. 2. Which of the following elements is most similar in behavior to calcium? a. magnesium b. sodium c. sulfur d. Chlorine 3. The electron configurations of main-group elements end in a. d and f orbitals. b.

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Section 5.4 Review 27. Describe how the periodic table was developed. 28. What criteria did Mendeleev use to construct his periodic table of elements? 29. Relate group, period and transition metals to the periodic table. 30. Identify each element as a metal metalloid, or nonmetal. a. gold b. silicon c. manganese d. sulfur e. barium 31.

Chapter 5 Atomic Structure and the Periodic Table

Chapter 5 Review Key Terms. ampere (amp) SI unit for current; circuit complete path that an electrical current travels along. conventional current current that flows through a circuit from the positive terminal of a battery through the circuit to the negative terminal of the battery

Chapter 5 Review - Introduction to Electricity, Magnetism ...

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